

A *p*-cymene-ruthenium(II)–DMSO complex, [(η^6 -C₁₀H₁₄)RuCl₂(DMSO)]

Manish Chandra,^a D. Shankar Pandey,^a M. Carmen Puerta^b and Pedro Valerga^{b*}

^aChemistry Department, APS University, Rewa 486 003, India, and ^bDepartamento de Ciencia de Materiales e Ingeniería Metalúrgica y Química Inorgánica, Facultad de Ciencias, Universidad de Cádiz, 11510 Puerto Real, Cádiz, Spain

Correspondence e-mail: pedro.valerga@uca.es

Key indicators

Single-crystal X-ray study
T = 293 K
Mean σ (C–C) = 0.006 Å
R factor = 0.033
wR factor = 0.090
Data-to-parameter ratio = 17.1

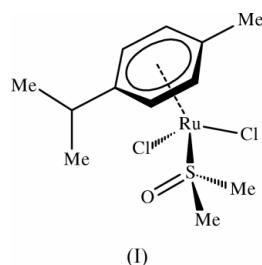
For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

The crystal structure of the neutral Ru^{II} complex [RuCl₂(η^6 -C₁₀H₁₄)(C₂H₆OS)] is reported. The overall coordination geometry about the ruthenium centre is best described as a typical ‘piano stool’. In the *p*-cymene ligand, the six C atoms of the arene ring are in a planar configuration, as expected.

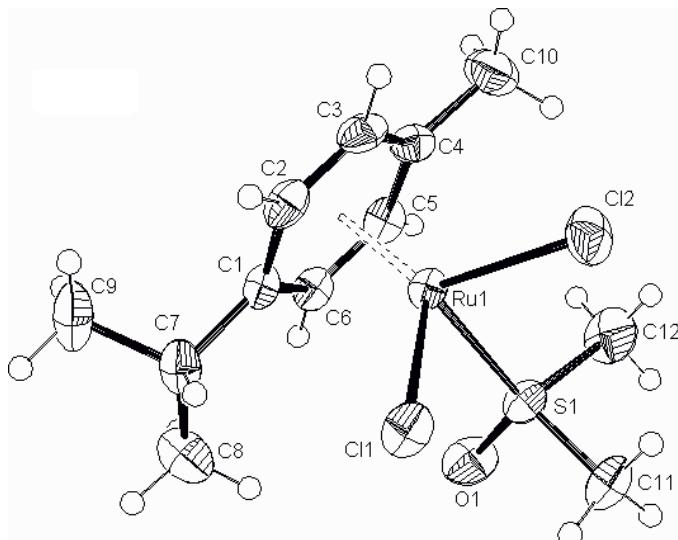
Received 4 December 2001
Accepted 10 December 2001
Online 14 December 2001

Comment

The homochiral sulfoxides have received considerable attention recently owing to their prominent role as intermediates in enantioselective synthesis and their potential use as precursors (Schenk *et al.*, 1997; Otto *et al.*, 2001). Because of our continuing interest in the area of arene ruthenium chemistry (Singh *et al.*, 2000), we reacted [(η^6 -C₁₀H₁₄)RuCl₂] with 4-cyanophenyl imidazole, which contained dimethylsulfoxide as solvate. Surprisingly, we isolated the title complex, [(η^6 -C₁₀H₁₄)RuCl₂(DMSO)], (I), instead of the expected arene–ruthenium–organonitrile complex.



The pertinent bond lengths and bond angles may be compared with the closely related 1,4,9,10-tetrahydroanthracene ruthenium complex [(η^6 -C₁₄H₁₄)RuCl₂(DMSO)] (Beasley *et al.*, 1993). The title compound adopts a typical piano-stool geometry (a description commonly used for half-sandwich compounds) with a pseudo-tetrahedral arrangement of the *p*-cymene, the two chloride ligands, and the S atom of the coordinated DMSO ligand around the ruthenium metal centre. This is a similar arrangement to that of [(η^6 -C₁₀H₁₄)RuCl₂(MePPh₂)], [(η^6 -C₁₀H₁₄)RuCl₂(pz)₂]PF₆ and the η^6 -arene DMSO complex [(η^6 -C₁₄H₁₄)RuCl₂(DMSO)] (Bennett *et al.*, 1972; Tocher *et al.*, 1983; Beasley *et al.*, 1993). The six-membered ring of the *p*-cymene is planar, with a maximum deviation of 0.013 (3) Å (C4) and an overall r.m.s. deviation of 0.008 Å. The C–C bond lengths within the *p*-cymene ring are similar, apart from a shortening of the C2–C3 bond. The Ru–C distances are almost equal, with an average of 2.199 (11) Å. The distance between the centroid of the *p*-cymene ring and ruthenium is 1.683 (3) Å, which is very close to that reported in other Ru^{II} arene complexes (Watkins

**Figure 1**

The structure of $[(\eta^6\text{-C}_{10}\text{H}_{14})\text{RuCl}_2(\text{DMSO})]$ showing 50% probability displacement ellipsoids.

& Fronczek, 1982; McCormick *et al.*, 1993; Gupta *et al.*, 1998). The Ru—Cl1 and Ru—Cl2 bond lengths are similar to those in other Ru^{II} complexes (Bruce *et al.*, 1981). The DMSO ligand is bonded to ruthenium *via* the S atom, resulting in a roughly tetrahedral arrangement around sulfur. The Ru—S distance is comparable to those observed in the η^6 -arene dimethyl sulf-oxide complex $[(\eta^6\text{-C}_{10}\text{H}_{14})\text{RuCl}_2(\text{DMSO})]$ (Beasley *et al.*, 1993) and the Ru^{II} complex $[\text{RuCl}_2(\text{DMSO})_4]$ (Mercer & Trotter, 1975).

Experimental

The complex $[(\eta^6\text{-C}_{10}\text{H}_{14})\text{RuCl}_2(\text{DMSO})]$ was prepared in a 68% yield by reaction of $[(\eta^6\text{-C}_{10}\text{H}_{14})\text{RuCl}_2]_2$ (0.612 g, 1 mmol) in dichloromethane (25 ml) with 4-cyanophenyl imidazole (0.338 g, 2 mmol) which was previously isolated from dimethyl sulfoxide and therefore contained DMSO as solvate. Orange-red crystals suitable for single-crystal analysis were obtained by slow diffusion of petroleum ether (b.p. 333–353 K) into a solution of the complex in dichloromethane.

Crystal data

$[\text{RuCl}_2(\text{C}_{10}\text{H}_{14})(\text{C}_2\text{H}_6\text{OS})]$
 $M_r = 384.31$
Triclinic, $P\bar{1}$
 $a = 7.8522(16)$ Å
 $b = 9.4124(9)$ Å
 $c = 10.9151(17)$ Å
 $\alpha = 106.850(11)^\circ$
 $\beta = 94.395(15)^\circ$
 $\gamma = 99.697(13)^\circ$
 $V = 754.3(2)$ Å³

$Z = 2$
 $D_x = 1.692 \text{ Mg m}^{-3}$
Mo $K\alpha$ radiation
Cell parameters from 25 reflections
 $\theta = 5.2\text{--}12.1^\circ$
 $\mu = 1.51 \text{ mm}^{-1}$
 $T = 293(2)$ K
Needle, brown
0.4 × 0.1 × 0.1 mm

Data collection

Enraf–Nonius CAD-4 diffractometer
 $\omega/2\theta$ scans
Absorption correction: ψ scan (North *et al.*, 1968)
 $T_{\min} = 0.836$, $T_{\max} = 0.860$
2632 measured reflections
2632 independent reflections

2373 reflections with $I > 2\sigma(I)$
 $\theta_{\max} = 25.0^\circ$
 $h = -9 \rightarrow 9$
 $k = 0 \rightarrow 11$
 $l = -12 \rightarrow 12$
3 standard reflections every 100 reflections
intensity decay: none

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.033$
 $wR(F^2) = 0.090$
 $S = 1.07$
2632 reflections
154 parameters

H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.067P)^2]$
where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} = 0.001$
 $\Delta\rho_{\max} = 1.26 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\min} = -0.83 \text{ e } \text{\AA}^{-3}$

Table 1
Selected geometric parameters (Å, °).

| | | | |
|-------------|------------|------------|------------|
| Ru1—Cl1 | 2.4111(10) | Ru1—C4 | 2.205(4) |
| Ru1—Cl2 | 2.4045(10) | Ru1—C5 | 2.192(3) |
| Ru1—S1 | 2.3400(10) | Ru1—C6 | 2.181(3) |
| Ru1—C1 | 2.201(3) | S1—O1 | 1.477(3) |
| Ru1—C2 | 2.207(4) | S1—C11 | 1.774(4) |
| Ru1—C3 | 2.209(3) | S1—C12 | 1.784(4) |
| Cl1—Ru1—Cl2 | 88.05(4) | C12—S1—Ru1 | 109.49(16) |
| S1—Ru1—Cl1 | 86.28(4) | O1—S1—C11 | 108.2(2) |
| S1—Ru1—Cl2 | 86.74(4) | O1—S1—C12 | 107.5(2) |
| O1—S1—Ru1 | 115.71(13) | C11—S1—C12 | 99.5(2) |
| C11—S1—Ru1 | 115.07(15) | | |

The highest residual peak lies 0.919 Å from Ru1. All H atoms were placed in calculated positions and refined with a riding model.

Data collection: CAD-4 Software (Enraf–Nonius, 1989); cell refinement: CAD-4 Software; data reduction: maxus (Mackay *et al.*, 1998); program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEP-3 (Farrugia, 1997); software used to prepare material for publication: SHELXL97.

Support of this work by the Council of Scientific and Industrial Research [HRDG, 01(1587)/99/EMR-II] is gratefully acknowledged.

References

- Beasley, T. J., Brost, R. D., Chu, C. K., Grundy, S. L. & Stobart, S. R. (1993). *Organometallics*, **12**, 4599–4606.
- Bennett, M. A., Robertson, G. B. & Smith, A. K. (1972). *J. Organomet. Chem.* **41**, C41–43.
- Bruce, M. I., Wong, F. S., Skelton, B. W. & White, A. H. (1981). *J. Chem. Soc. Dalton Trans.* pp. 1398–1405.
- Enraf–Nonius (1989). CAD-4 Software. Version 5.0. Enraf–Nonius, Delft, The Netherlands.
- Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.
- Gupta, D. K., Sahay, A. N., Pandey, D. S., Jha, N. K., Sharma, P., Espinosa, G., Cabrera, A., Puerta, M. C. & Valerga, P. (1998). *J. Organomet. Chem.* **568**, 13–20.
- Mackay, S., Gilmore, C. J., Edwards, C., Tremayne, M., Stuart, N. & Shankland, K. (1998). maxus. University of Glasgow, Scotland, Nonius BV, Delft, The Netherlands, and MacScience Co. Ltd, Yokohama, Japan.
- McCormick, F. B., Cox, D. D. & Gleason, W. B. (1993). *Organometallics*, **12**, 610–612.
- Mercer, A. & Trotter, J. (1975). *J. Chem. Soc. Dalton Trans.* pp. 2480–2483.
- North, A. C. T., Phillips, D. C. & Mathews, F. S. (1968). *Acta Cryst. A* **24**, 351–359.
- Otto, M., Boone, B. J., Arif, A. M. & Gladysz, J. A. (2001). *J. Chem. Soc. Dalton Trans.* pp. 1218–1229.
- Schenk, W. A., Frisch, J., Durr, M., Burzlaff, N., Stalke, D., Fleischer, R., Adam, W., Precht, F. & Smerz, A. K. (1997). *Inorg. Chem.* **36**, 2372–2378.
- Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
- Singh, A., Sahay, A. N., Pandey, D. S., Puerta, M. C. & Valerga, P. (2000). *J. Organomet. Chem.* **605**, 74–81, and references therein.
- Tocher, D. A., Gould, R. O., Stephenson, T. A., Bennett, M. A., Ennett, J. P., Matheson, T. W., Sawyer, L. & Shah, V. K. (1983). *J. Chem. Soc. Dalton Trans.* pp. 1671–1673.
- Watkins, S. F. & Fronczek, F. R. (1982). *Acta Cryst. B* **38**, 270–271.